

A New Method for the Preparation of the 6,6 and 6,7 Ring Systems of Bicyclic Ethers

Fabienne Simart, Yves Brunel, Sylvie Robin, Gérard Rousseau*

Laboratoire des Carbocycles (Associé au CNRS), Institut de Chimie Moléculaire d'Orsay Bât. 420, Université de Paris-Sud, 91405 Orsay (France)

Received 14 April 1998; accepted 3 September 1998

Abstract: Preparation of 2,7-dioxabicyclo[4.4.0]decane and 2,8-dioxabicyclo[5.4.0]undecane derivatives in five steps from dihydropyran are reported. Transformation of 3-bromo-2-allyloxepane into 2,7-dioxabicyclo-[5.4.0]-undecane is also described. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: Cyclization; oxepanes; halogens and compounds; enol ethers.

INTRODUCTION

An always increasing number of natural products¹ possessing polycyclic ether frameworks, such as ciguatoxins, maitotoxin or brevetoxins, have inspired numerous research groups to study original ways to synthesize them.^{2,3} These efforts have recently culminated in the synthesis of some of these natural products.⁴

Preparation of tetrahydropyran derivatives by 6-endo electrophilic cyclizations of hydroxy-enol ethers has been reported to lead to interesting synthetic applications.⁵ Preparation of oxepane derivatives by similar 7-endo electrophilic cyclizations does not seem to be know. However, we recently reported that this kind of cyclization could be carried out using bis(sym-collidine)iodine(I) hexafluorophosphate as an electrophilic reagent (scheme 1).⁶ We decided to examine if the resulting 2,3-difunctionalized ethers could be transformed into bicyclic ethers with the same functionalities, thus allowing an iterative construction of polycyclic ethers (Scheme 1).

^{*} E-mail: groussea@icmo.u-psud.fr; Fax: 33 169156278

RESULTS

Preparation of 6,6 bicyclic ethers

Epoxidation of dihydropyran in methanol led to 2-hydroxy-1-methoxytetrahydropyran in good yield $(70\%)^7$ which was readily alkylated at the 2 position using allyltrimethylsilane in the presence of boron-trifluoride diethyletherate in acetonitrile $(73\% \text{ yield}).^8$ 2-Allyl-3-hydroxytetrahydropyran **2** was obtained as a 65: 35 *cis-trans* mixture of diastereoisomers (Scheme 2).

Scheme 2

Protection of the alcohol as the acetate or *tert*-butyldimethylsilylether before the reaction with allyltrimethylsilane did not improve the selectivity. However, the two diastereoisomers could be easily separated by liquid chromatography over silica gel and were fully characterized from their ¹H and ¹³C-NMR spectra. The subsequent reactions were conducted on each pure diastereoisomer. Ozonolysis of *trans* pyran 2a led to the hydroxyaldehyde 3a characterized from its NMR and IR spectra. This compound reacted with methoxymethyltriphenylphosphorane to afford the enol ether 4a, as a *Z:E* mixture (20:80). Reaction of this mixture with bis(*sym*-collidine)iodine(I) hexafluorophosphate led to the bicyclic iodo ether 5a in moderate yield (50%) (Scheme 3), as a mixture of the four diastereoisomers (14:21:26:39), whose relative configurations could not be assigned.

Scheme 3

The same reaction sequence was conducted in the *cis* serie. In this case, the ozonolysis led to furanol **3b** which after Wittig reaction and iodoetherification led to a mixture of three diastereoisomers (15:15:70) of the bicyclic compound in good yield (82%), whose relative configurations could not be assigned (Scheme 3).

Preparation of 6,7 bicyclic ethers.

This strategy was applied to the preparation of the 6,7 bicyclo compounds. Hydrolysis of enol ether 4a using perchloric acid led to the corresponding hemiacetal 6a which after Wittig reaction using methoxymethyltriphenylphosphorane, was transformed into the enol ether 7a. The iodoetherification of this latter carried out in methylene chloride in the presence of bis(sym-collidine)iodine(I) hexafluorophosphate giving the desired bicyclic compound 8a in moderate yield (38%) as a mixture of three diastereoisomers (Scheme 3). These products were characterised from their NMR and mass spectra.

Scheme 4

The preparation of the *cis* 6,7 bicyclic compound **8b** was performed in an analogous manner starting from the enol ether **4b**. Hydrolysis, followed by the Wittig reaction and iodoetherification led to the *cis* bicyclo ether **8b** as a mixture of three diastereoisomers (Scheme 4). As in the 6,6 series, it appreared that the iodoetherification of the *cis* compound was more efficient than the *trans* series.

Preparation of 7,6 bicyclic ether.

In view of the preparation of polycyclic ethers analogous to brevetoxin derivatives, we decided to test the preparation of bicyclic compounds starting from oxepane 1. While the allylation of tetrahydropyran is well known,⁸ to our knowledge nothing has been reported starting with oxepanes. Reaction of oxepane 1 with allyltrimethylsilane in the presence of Lewis acids such as AlCl₃, ZnCl₂, SnCl₄, BF₃·Et₂O, or CF₃SO₂SiMe₃ led to degradation products and (or) oxepane ring cleavage products. After reaction with 4-nitrobenzyl alcohol in the the presence of *p*-toluenesulfonic acid and a radical scavenger, the resulting acetal 9 was C-2 alkylated with allyltrimethylsilane in the presence of BF₃·Et₂O. An inseparable mixture of 2-allyloxepane 10 and 2-allyl-3-iodooxepane 11 was obtained (Scheme 5). In the absence of the radical inhibitor, 2-allyloxepane 10 was obtained as the sole product (57%).

Scheme 5

To avoid radical halogen removal, the more stable bromine derivative 13 was studied. This substrate was prepared by a 7-endo cyclization of alcohol 12 using bis(sym-collidine)bromine(I) hexafluorophosphate (74% yield) (Scheme 6).

OH OMe
$$Br^+(NO)$$
 PF_6 OMe Br
 CH_2Cl_2 Br

Scheme 6

As with the iodine derivative 1, the transacetalization in the presence of 4-nitrobenzyl alcohol led to the desired acetal 14 which was alkylated using allyltrimethylsilane in the presence of BF₃·Et₂O without bromine removal (74%). Only one diastereoisomer, 15, of unknown stereochemistry was isolated. Reaction of the bromooxepane 15 with 2 equivalents of silver trifluoroacetate in a mixture of water-nitromethane at pH 7 led to the corresponding alcohol 16 as a 97:3 mixture (87.5% yield). The stereochemistry of the main isomer was found to be *trans* by analysis of its ¹H-NMR spectra. Indeed, by irradiation of the CH₂ of the allyl group a coupling constant of 7 Hz was found between the two hydrogens H_a and H_b (Scheme 7). This stereochemistry was confirmed by calculation of the conformation of each diastereoisomers by molecular modelling (Pro Chemist Model 5.3 and MAD programs).

The carbon-carbon double bond of this *trans* alcohol was cleaved with ozone and the resulting hydroxyaldehyde 17 reacted with methoxymethyltriphenylphosphorane to provide the enol ether 18 (45% yield from 16). The bromoetherification in the presence of bis(*sym*-collidine)bromine(I) hexafluorophosphate gave the desired bicyclic compound 19 (31% yield) as a mixture of three diastereoisomers (Scheme 7).

O
$$OCH_2PhpNO_2$$

Br OCH_2PhpNO_2

Br OCH_2PhpNO_2

Br OCH_2PhpNO_2
 OCH_2Cl_2
 OCH_2Cl_2

In conclusion, these results have shown that it is possible to prepare by haloetherification, 6,6-, 6,7- and 7,6 bicyclic ethers with cis or trans stereochemistry present in natural products such as maitotoxin. Work is now in progress to improve these transformations and to apply this reaction pathway to an iterative preparation of polycyclic ethers.

EXPERIMENTAL

¹H and ¹³C spectra were recorded in CDCl₃ solution on Brucker AM 200, 250 and 50.3, 62.9 MHz, respectively. Infrared spectra were recorded on a Perkin-Elmer 682. Mass spectra were determined on a NERMAG R₁₀ - 10 spectrometer. Flash chromatography was performed on SiO₂ (SDS 35 - 70 microns). Bis(sym-collidine)iodine(I) hexafluorophosphate and bis(sym-collidine)bromine(I) hexafluorophosphate are abbreviated as HBI and HBB respectively. HBB was prepared as reported for HBI.⁶

3-Iodo-2-methoxyoxepane 1. To 6-methoxyhex-5-en-1-ol (1 mmol) in dichloromethane (20 mL) was added HBI (1.2 equiv., 1.2 mmol, 620 mg) at rt. After half an hour, silica gel (3 g) was added and solvent was evaporated under vacuum. Purification of the product by column chromatography (ether:hexane; 5:95) gave 215 mg (0.84 mmol) of a mixture of two diastereoisomers (*cis:trans*; 35:65, 84% yield). *Trans*-diastereoisomer **1a**. ¹H NMR δ 1.35 - 1.80 (m, 4H), 2.06 - 2.37 (m, 2H), 3.37 (s, 3H), 3.54 (ddt, J = 12, 2.6, 1 Hz, 1H), 3.82 (m, 1H), 4.08 (ddd, J = 10.5, 7.8, 2.6 Hz, 1H), 4.85 (d, J = 7.8 Hz, 1H). ¹³C NMR δ 28.9, 29.6, 32.5, 37.4, 55.6, 63.2, 109.5. MS EI (m/z, %): 61(81), 69 (100), 97 (37), 129 (37), 154 (30). *Cis*-diastereoisomer **1b**. ¹H NMR δ 1.47 - 1.95 (m, 4H), 2.02-2.35 (m, 2H), 3.45 (s, 3H), 3.72 (m, 1H), 3.94 (m, 1H), 4.19 (d, J = 2 Hz, 1H), 4.33 (ddd, J = 5.2, 4.2, 1 Hz, 1H). ¹³C NMR δ 24.7, 29.5, 31.8, 35.8, 55.9, 64.8, 103.3. MS EI (m/z, %): 61(100), 69 (86), 71(34), 97 (28), 129 (26), 154 (23).

2-Allyl-3-hydroxytetrahydropyran 2. To a solution of 3-hydroxy-2-methoxytetrahydropyran (50.7 mmol) in acetonitrile (530 mL), under argon were added at 0°C, allyltrimethylsilane (2.5 equiv., 127 mmol, 20 mL), and boron trifluoride diethyletherate (2.5 equiv., 127 mmol, 16 mL). The solution was allowed to warm to rt and stirred for 3 h. Saturated aqueous sodium hydrogen carbonate was added and the organic layer extracted with ether. The combined organic extracts were dried (Na₂SO₄), concentrated and purified by flash silica gel column chromatography (ether:hexane; 60:40). A mixture of two diastereoisomers partially separated was obtained (5.2 g, *cis:trans*; 65:35, 73% yield). *Trans*-diastereoisomer **2a**. H NMR δ 1.44 - 1.6 (m, 2H), 1.60 (s, 1H, OH), 1.68 (m, 1H), 2.11 (m, 1H), 2.30 (m, 1H), 2.60 (m, 1H), 3.12 (td, J = 7.5, 4 Hz, 1H), 3.28 - 3.43 (m, 2H), 3.92 (m, 1H), 5.13 (m, 2H), 5.94 (m, 1H). *Cis*-diastereoisomer **2b**. H NMR δ 1.30 - 2.04 (m, 4H), 2.32 (m, 2H), 3.37 (m, 1H), 3.70 (m, 2H), 4.00 (m, 1H), 5.11 (m, 2H), 5.80 (m, 1H), 8.02 (s, 1H, OH). MS EI (*m/z*, %): 41(30), 43 (54), 44 (49), 55 (17), 57 (16), 71 (21), 101 (100), 142 (1). IR (film) (cm⁻¹): 3440, 3080, 2950, 2860, 1650, 1440, 1100, 1000.

Ozonolysis of olefins 2a and 2b. Ozone-oxygen gas was bubbled into a solution of 2-allyl-3hydroxytetrahydropyran 2 (10.5 mmol, 1.5 g) in a 4:1 dichloromethane-methanol mixture (70 mL) cooled at -78°C, until a persistent blue color appeared. The solution was stirred for 30 min at -78°C and argon was bubbled into the solution to eliminate the excess ozone. Triphenylphosphine (15.75 mmol, 4.15 g, 1.5 equiv.) was added and the mixture was allowed to warm to rt. The solution was concentrated under vacuum and the crude product purified by silica gel column chromatography (ether:hexane; 60:40) to give 8-hydroxy-2,7dioxabicyclo(4.3.0)nonane 3b (1.15 g, 76% yield) from cis-isomer 2b and 2-(2-oxoethyl)-3hydroxytetrahydropyran 3a (1.03 g, 68% yield) from trans-isomer 2a. (2S*,3R*)-2-(2-oxoethyl)-3hydroxytetrahydropyran 3a. ¹H NMR δ 1.32 - 2.24 (m, 4H), 2.31 - 2.45 (s, 1H, OH), 2.55 (ddd, J = 16, 8, 2 Hz. 1H), 2.90 (ddd, J = 16, 5, 2 Hz, 1H), 3.26 - 3.45 (m, 2H), 3.60 (ddd, J = 7.5, 5.3, 3.3 Hz, 1H), 3.88 (m, 1H), 9.77 (t, J = 2 Hz, 1H). ¹³C NMR δ 25.3, 32.8, 46.7, 67.6, 69.8, 77.6, 201.9. Anal. Calcd for C7H₁₂O₃: C. 58.30 ; H. 8.39. Found: C, 58.40; H, 8.18. (1R*,6R*)-8-hydroxy-2,7-dioxabicyclo[4.3.0]nonane 3b. ¹H NMR $\delta 1.32 - 1.54$ (m, 1H), 1.63 - 1.82 (m, 2H), 1.85 - 2.12 (ddt, J = 15, 4.2, 5 Hz, 2H), 2.09 - 2.23 (m, 1H), $2.30 \text{ (dd, J} = 15, 5 \text{ Hz, } 0.6\text{H, OH} \ \alpha \text{ isomer)}, 2.95 \text{ (d, J} = 3.4 \text{ Hz, } 0.4\text{H, OH} \ \beta \text{ isomer)}, 3.27 - 3.44 \text{ (ddt, J} = 2, 12)$ Hz, 1H), 3.85 (m, 1H), 3.92 - 4.15 (m, 2H), 5.35 (dt, J = 3.4, 12.7 Hz, 0.4H, β isomer), 5.75 (dt, J = 3.4, 5 Hz, 0.6H, α isomer). ¹³C NMR δ α isomer: 20.1, 25.6, 42.0, 66.8, 75.8, 76.7, 99.1. β isomer: 20.1, 25.6, 42.8, 66.4, 74.2, 77.0, 98.4. MS EI (m/z, %): 41 (34), 42 (22), 43 (56), 44 (37), 45 (21), 55 (31), 56 (27), 57 (37), 71

(100), 73 (30), 80 (20), 83 (23), 84 (18), 98 (20), 126 (21), 144 (2). IR (film) (cm⁻¹): 3420, 2960, 2860, 1450, 1240, 1130, 1100, 1060, 1020. Anal. Calcd for C7H₁₂O₃: C, 58.30; H, 8.39. Found: C, 58.47; H, 8.12.

3-Hydroxy-2-(3-methoxyprop-2-enyl)tetrahydropyran 4. To a mixture of methoxymethyltriphenylphosphonium chloride (20.8 mmol, 7.15 g, 3 equiv.) and anhydrous THF (52 mL) under argon atmosphere at -78°C, was added dropwise tertbutyllithium (20.8 mmol, 13.9 mL of a 1.5 M soln. in pentane, 3 equiv.). After 10 min. of stirring, a solution of lactol 3a (or aldehyde 3b) (6.9 mmol) in THF (10 mL) was added. The mixture was allowed to warm to rt and stirred for 15 h. The reaction was quenched by addition of water (10 mL). The solvent was evaporated under vacuum and the aqueous layer extracted with methylene chloride. The combined organic layers were dried (MgSO₄), filtered and concentrated under vacuum. The crude product was purified by flash chromatography on silica gel (ether:hexane; 80:20) to obtain a mixture of two isomers (Z:E; 45:55, 725 mg, 73% yield for 4a), (Z:E; 35:65, 937 mg, 79% yield for 4b). (2S*,3R*)-3-Hydroxy-2-(3methoxyprop-2-enyl)tetrahydropyran 4a. ¹H NMR δ 1.32 - 2.13 (m, 4H), 2.16 (t, J = 6.8 Hz, 0.8H, OH), 2.25 - 2.70 (m, 2H), 2.94 (m, 0.2H, OH), 2.95 - 3.2 (m, 1H), 3.25 - 3.45 (m, 2H), 3.53 (s, 2.4H, OMe), 3.62 (s, 0.6H, OMe), 3.80 - 3.85 (m, 1H), 4.53 (dt, J = 8.4, 6.3 Hz, 0.2H), 4.82 (dt, J = 12.6, 7.3 Hz, 0.8H), 6.03 (d, J = 1.0H), 6.036.3 Hz, 0.2H), 6.38 (d, J = 12.6 Hz, 0.8H). ¹³C NMR δ E-isomer : 25.4, 30.2, 32.6, 55.6, 67.4, 69.4, 82.4, 98.3, 148.1. Z-isomer: 25.5, 26.1, 31.4, 59.4, 67.6, 68.9, 82, 101.9, 147.4. Anal. Calcd for C9H₁₆O₃: C, 62.75; H, 9.37. Found: C, 62.86; H, 9.35. (2R*,3R*)-3-Hydroxy-2-(3-methoxyprop-2-enyl)tetrahydropyran **4b.** 1 H NMR δ 1.25 -1 .45 (m, 1H), 1.50 - 1.17 (m, 1H), 1.78 - 2.20 (m, 2H), 2.10 - 2.48 (m, 2H+OH Eisomer), 2.68 (d, J = 6.8 Hz, 0.65H, OH Z-isomer), 3.15 - 3.3 (m, 2H), 3.46 (dt, J = 11, 1 Hz, 1H), 3.49 (s, 1.95H, OMe E-isomer), 3.59 (s, 1.05H, OMe Z-isomer), 3.60 - 3.71 (m, 1H), 3.9 - 4.1 (m, 1H), 4.38 (dt, J =9.5, 5.8 Hz, 0.35H), 4.69 (dt, J = 7.3, 13.2 Hz, 0.65H), 5.98 (d, J = 5.8 Hz, 0.35H), 6.35 (d, J = 13.2 Hz, 0.65H). ¹³C NMR δE -isomer: 20.0, 29.9, 30.5, 55.8, 65.4, 68.5, 80.5, 98.0, 148.7. Z-isomer: 20.1, 26.3, 30.0, 59.5, 65.2, 68.4, 79.0, 101.6, 147.9. MS EI (m/z, %): 41 (39), 43 (82), 55 (44), 59 (19), 69 (15), 71 (74), 72 (25), 83 (29), 84 (16), 100 (14), 101 (100). IR (film) (cm⁻¹): 3450, 3060, 2960, 2860, 1660, 1450, 1220, 1090, 1000. Anal. Calcd for C9H₁₆O₃: C, 62.75; H, 9.37. Found: C, 62.80; H, 9.30.

4-Iodo-3-methoxy-2,7-dioxabicyclo[4.4.0]decane 5. To a mixture of enol ether 4a (or 4b) (2.9 mmol, 500 mg) in methylene chloride (60 mL), HBI (3.5 mmol, 1.79 g, 1.2 equiv.) was added at room temperature. The mixture was stirred for 15 min. Silica gel (500 mg) was added to the solution and the solvent removed under vacuum. Purification by chromatography on silica gel column (ether:hexane; 30:70) led to a mixture of three diastereoisomers (15:15:70) from which only one isomer could be separated for compound 5b (2.4 mmol, 82% yield) and to four partially separable diastereoisomers (14:21:26:39) of compound 5a (1.5 mmol, 50%) yield). (1R*, 6S*)-4-iodo-3-methoxy-2,7-dioxabicyclo[4.4.0]decane 5a, 2 isomers, C8-C9 trans:cis; 90:10. ¹H NMR δ 1.43 - 1.8 (m, 3H), 2.00 - 2.11 (m, 1H), 2.11 (dt, J = 12.8, 11.2 Hz, 1H), 2.27 - 2.40 (m, 0.2H), $2.63 \text{ (dt, } J = 12.6, 4.5 \text{ Hz, } 1\text{H}), 2.96 \text{ (ddd, } J = 10.8, 8.4, 4.5 \text{ Hz, } 1\text{H}), 3.17 \text{ (ddd, } J = 8.9, 8.4, 4.2 \text{ Hz, } 1\text{H}), 3.28 \text{ (dt, } J = 12.6, 4.5 \text{ Hz, } 1\text{H}), 2.96 \text{ (ddd, } J = 10.8, 8.4, 4.5 \text{ Hz, } 1\text{H}), 3.17 \text{ (ddd, } J = 8.9, 8.4, 4.2 \text{ Hz, } 1\text{H}), 3.28 \text{ (dt, } J = 12.6, 4.5 \text{ Hz, } 1\text{H}), 2.96 \text{ (ddd, } J = 10.8, 8.4, 4.5 \text{ Hz, } 1\text{H}), 3.17 \text{ (ddd, } J = 8.9, 8.4, 4.2 \text{ Hz, } 1\text{H}), 3.28 \text{ (dt, } J = 12.6, 4.5 \text{ Hz, } 1\text{Hz, } 1\text$ - 3.40 (m, 1H), 3.41 (s, 0.3H, OMe cis-isomer), 3.50 (2.7H, OMe trans-isomer), 3.85 - 4.00 (m, 1.9H), 4.05 -4.15 (m, 0.1H), 4.43 (d, J = 8.8 Hz, 1H), 4.63 (d, J = 3.9 Hz, 0.1H). ¹³C NMR δ 21.4/24.03, 24.95/25.46, 28.9/29.0. 36.9/41.9. 55.5/57.3, 67.0/67.8, 68.2/75.6, 77.7/78.3, 99.1/105.6. 2 isomers, C8-C9 trans:cis; 35:65. $1_{\text{H NMR }\delta}$ 1.30 - 1.84 (m, 3H), 1.88 - 2.40 (m, 1H), 2.04 - 2.15 (m, 1.3H), 2.25 - 2.40 (m, 0.7H), 3.00 (ddd, J) = 9.9, 7.9, 5.9 Hz, 0.35H), 3.29 - 3.60 (m, 2.65H), 3.50 (s, 1.95H, OMe cis-isomer), 3.41 (s, 1.05H, OMe trans-isomer), 3.83 - 3.97 (m, 1H), 4.04 - 4.15 (m, 0.35H), 4.34 (t, J = 2.9 Hz, 0.65H), 4.63 (d, J = 3.9 Hz, 0.35H), 4.85 (s, 0.65H). ¹³C NMR δ 21.4/25.8, 25.5/26.2, 28.8/28.9, 33.8/37.0, 54.6/55.5, 67.9/68.1, 68.2/69.7, 75.1/78.3, 99.1/101.5. Anal. Calcd for C9H₁₅O₃I: C, 36.24; H, 5.07. Found: C, 36.23; H, 5.10. (1R*,6R*)-4-iodo-3-methoxy-2,7-dioxabicyclo[4.4.0]decane **5b**, 2 isomers, C3-C4 trans:cis; 50:50. ¹H NMR δ 1.15 - 1.43 (m, 1H), 1.55 - 1.78 (m, 1H), 1.86 - 2.09 (m, 2H), 2.14 - 2.31 (m, 1H), 2.49 - 2.65 (m, 1H), 3.15 - 1.43 (m, 1H), 2.49 - 2.65 (m, 1H), 3.15 - 1.43 (m, 1H), 3.15 (m, 1H 3.22 (m, 0.5H), 3.29 - 3.48 (m, 1H), 3.36 (s, 1.5H), 3.50 (s; 1.5H), 3.53 - 3.68 (m, 1H), 3.75 - 3.82 (m, 0.5H), 3.88 - 4.10 (m, 1.5H), 4.11 - 4.28 (m, 0.5H), 4.40 (d, J = 7.9 Hz, 0.5H), 5.05 (s, 0.5H). 13 C NMR δ 16.7/20.1, 20.5/25.0, 28.0/28.1, 32.7/42.7, 54.7/56.5, 62.5/67.8, 67.4/71.3, 70.2/74.0, 103.1/105.5. 1 isomer. 1 H NMR $_{\delta}$ 1.30 (dq, J = 13 Hz, 1H), 1.62 (ddd, J = 14.4, 4.9, 3.9 Hz, 1H), 1.76 - 2.05 (m, 2H), 2.24 (dt, J = 13.6, 3.5 Hz, 1H), 2.50 (td, J = 3.2, 13.4 Hz, 1H), 3.30 (t, J = 2.9 Hz, 1H), 3.40 (s, 3H), 3.40 (td, J = 11.8, 1.9 Hz, 1H), 3.80 (s, 1H), 3.97 (m, 1H), 4.49 (ddd, J = 13.1, 4.6, 3 Hz, 1H), 4.72 (d, J = 3 Hz, 1H). 13 C NMR $_{\delta}$ 20.2, 22.4, 27.9, 36.9, 55.6, 62.4, 67.8, 74.4, 99.9. MS EI (m/z, %): 41 (42), 43 (29), 55 (75), 71 (100), 84 (31), 111 (27), 139 (15), 140 (16), 171 (42), 298 (3). IR (film) (cm $^{-1}$): 2960, 2930, 2850, 1450, 1150, 1130, 1070, 1030. Anal. Calcd for C9H₁₅O₃I: C, 32.26; H, 5.07. Found: C, 32.49; H, 5.11.

3-Hydroxy-2,7-dioxabicyclo[4.4.0]decane 6. To a solution of enol ether **4a** (or **4b**) (2.9 mmol) in THF (10 mL) was added at rt, 5 mL of a 70% aqueous perchloric acid solution. The mixture was stirred for 1 h. The reaction was quenched by addition of a saturated sodium hydrogen carbonate solution. The aqueous layer was extracted with methylene chloride. The combined organic layers were dried (MgSO4), filtered and concentrated under vacuum. Chromatography of the crude product on a silica gel column (ether:hexane; 80:20) gave a mixture of two isomers (α:β; 50:50) which were not separated. (320 mg, 70% yield for **6a** from **4a**) (407 mg, 89% yield for **6b** from **4b**). (1R*,6S*)-3-hydroxy-2,7-dioxabicyclo[4.4.0]decane **6a**, 2 isomers α:β; 50:50.

¹H NMR δ 1.22 - 2.15 (m, 6H), 3.00 (m, 1H, OH), 3.14 (ddd, J = 4.7, 10.5, 8.5 Hz, 0.5H), 3.20 - 3.52 (m, 1.5H), 3.68 (m, 1H), 3.9 (m, 1.5H), 4.34 (m, 0.5H), 4.8 (m, 0.5H), 5.20 (s, 0.5H).

¹³C NMR δ 23.8/25.2, 25.5/25.7, 27.9/29.7, 29.2/32.1, 67.8/67.8, 69.1/75.2, 77.4/78.3, 90.9/95.9. (1R*,6R*)-3-hydroxy-2,7-dioxabicyclo[4.4.0]decane **6b**, 2 isomers α:β; 50:50.

¹H NMR δ 1.20 - 2.20 (m, 8H), 3.26 (s, 0.5H, OH), 3.30 - 3.56 (m, 2H), 3.83 (m, 0.5H, OH), 3.91 - 4.45 (m, 2H), 4.74 (m, 0.5H), 5.34 (s, 0.5H).

¹³C NMR δ 20.2/20.3, 23.1/23.9, 27.5/28.1, 28.4/28.6, 63/67.8, 67.9/70.7, 70.9/71.5, 91.7/96.4. MS E1 (*m/z*, %): 41 (15), 43 (25), 55 (15), 56 (15), 57 (12), 71 (100), 84 (10), 112 (11), 140 (7), 158 (5). IR (film) (cm⁻¹): 3400, 2950, 2860, 1450, 1340, 1270, 1240, 1200, 1140, 1120, 1100, 1050, 1010, 900.

3-Hydroxy-2-(4-methoxybut-5-enyl)tetrahydropyran 7. The reaction conditions were the same as for the preparation of compound 4a and 4b. From lactol 6a (or 6b) (1.58 mmol), after purification by silica gel column chromatography (ether: hexane; 80: 20), we obtained enol ether 7a (130 mg, 52% yield from 6a) and enol ether 7b (197 mg, 67% yield from 6b) as a mixture of two isomers (E:Z; 80:20) not separated. (2S*,3R*)-3-hydroxy-2-(4-methoxybut-5-enyl)tetrahydropyran **7a**. ¹H NMR δ 1.25 - 2.2 (m, 8H), 2.33 (s, 1H, OH), 2.98 (td, J = 9.5, 2.6 Hz, 1H), 3.26 (m, 2H), 3.47 (s, 2.4H, OMe), 3.54 (s, 0.6H, OMe), 3.85 (m, 1H), 4.34 (td, J = 6.3, 6.8 Hz, 0.2H), 4.70 (dt, J = 12.6, 7.3 Hz, 0.8H), 5.84 (d, J = 6.3 Hz, 0.2H), 6.28 (d, J = 12.6Hz, 0.8H). ¹³C NMR δ *E*-isomer 23.4, 25.5, 32.8, 33.0, 55.7, 67.3, 70.2, 81.4, 102.6, 147.1. Z-isomer 19.7, 25.3, 32.0, 32.7, 59.3, 67.5, 70.1, 81.8, 106.5, 146.1. (2R*,3R*)-3-hydroxy-2-(4-methoxybut-5-enyl)tetrahydropyran 7b. ¹H NMR δ 1.30 - 2.23 (m, 9H), 3.29 (dd, J = 10.5, 6.8 Hz, 1H), 3.40 (dd, J = 15.2, 3.7) Hz, 1H), 3.50 (s, 2.4H, OMe), 3.58 (s, 0.6H, OMe), 3.62 (m, 1H), 3.97 (m, 1H), 4.44 (q, J = 7.9 Hz, 0.2H), 4.70 (dt, J = 15.8, 9.5 Hz, 0.8H), 5.87 (d, J = 7.9 Hz, 0.2H), 6.40 (d, J = 15.8 Hz, 0.8H). 13 C NMR δE -isomer 20.2, 23.5, 30.6, 32.7, 55.8, 66.6, 68.4, 79, 102.2, 147.3. Z-isomer 19.8, 24.7, 27.2, 31.6, 59.4, 66.3, 67.2, 75.9, 101.3, 146.4. MS EI (m/z, %): 41 (32), 43 (27), 44 (25), 55 (19), 58 (23), 71 (100), 72 (17), 84 (50), 85 (19), 97 (20), 98 (26), 186 (3). IR (film) (cm⁻¹): 3440, 3020, 2950, 2860, 1660, 1450, 1210, 1160, 1130, 1100, 1060, 1000, 940. Anal. Calcd for C₁₀H₁₈O₃: C, 64.47; H, 9.75. Found: C, 64.50; H, 9.76.

Cyclization of enols ethers 7 with HBI. The reaction conditions were the same as for the preparation of compound 5a and 5b. From enol ether 7a (or 7b) (0.54 mmol) and after purification by silica gel column chromatography (ether:hexane; 10:90), we obtained (1R*,7S*)-4-iodo-3-methoxy-2,8-dioxabicyclo[5.4.0]-undecane 8a (64 mg, 38% yield from 7a) as a mixture of three diastereoisomers partially separated and (1R*,7R*)-4-iodo-3-methoxy-2,8-dioxabicyclo[5.4.0]undecane 8b (137 mg, 81% yield from 7b) as a mixture of three inseparable diastereoisomers. (1R*,7S*)-4-iodo-3-methoxy-2,8-dioxabicyclo[5.4.0]undecane 8a, C3-C4 trans-isomer. 1 H NMR δ 1.30 - 2.30 (m, 8H), 3.07 (dt, J = 5.5, 9.5 Hz, 1H), 3.3 (dt, J = 5.5, 10.5 Hz, 1H), 3.41 (s, 3H, OMe), 3.46 (m, 1H), 3.84 (m, 1H), 4.00 (ddd, J = 8.9, 6.3, 0.5 Hz, 1H), 4.90 (d, J = 8.9 Hz,1H).

¹³C NMR δ 25.3 30.4, 31.1, 33.4, 37.1, 56.2, 67.2, 68.7, 79.5, 108.4. C3-C4 (*cis:trans*; 50:50) isomers 1 H NMR δ 0.82 - 2.34 (m, 8H), 3.00 - 3.42 (m, 3H), 3.44 (s, 3H, OMe), 3.46 (s, 3H, OMe), 3.77 - 3.97 (m, 1.5H), 4.33 (m, 0.5H), 4.4 (s, 0.5H), 4.80 (d, J = 7.4 Hz, 1H). 13 C NMR δ 25.5/25.7, 29.5/29.6, 30.6/31.3, 32.7/34.0, 34.3/34.7, 56.1/56.6, 67.5/67.9, 67.9/68.0, 80.4/82.4, 100.3/112.9. Anal. Calcd for C₁₀H₁₇O₃I: C, 38.48; H, 5.49. Found: C, 38.60; H, 5.52. (1R*,7R*)-4-iodo-3-methoxy-2,8-dioxabicyclo[5.4.0]undecane **8b**, 3 isomers. 1 H NMR δ 1.27 - 2.17 (m, 7H), 2.40- 2.76 (m, 1H), 3.34 (s, 1.38H, OMe), 3.35 (s, 1.32H, OMe), 3.44 (s, 0.6H, OMe), 3.30 - 3.60 (m, 3H), 3.84 - 4.05 (m, 1.54H), 4.25 (td, J = 6.3, 4.73 Hz, 0.46H), 4.64 (d, J = 8.9 Hz, 0.44H), 4.69 (d, J = 3.2 Hz, 0.1H), 5.01 (d, J = 6.3 Hz, 0.46H). 13 C NMR δ 20.3/20.7/20.9, 29.0/29.5/29.6, 29.7/29.9/30, 30.5/32.4/33.3, 34.4/36/37, 56.2/56.6/56.4, 64.91/64.93/66.6, 67.6/67.9/68.5, 75.7/75.9/76, 101.1/108.3/112.5. MS EI (*m/z*, %): 41 (52), 42 (12), 43 (33), 55 (21), 58 (10), 71 (100), 72 (24), 97 (25), 98 (20), 125 (17); 153 (10), 185 (37), 252 (23), 312 (3). IR (film) (cm⁻¹): 2950, 2840, 1460, 1450, 1380, 1340, 1270, 1250, 1200, 1120, 1090, 1050, 1030, 960, 920, 900. Anal. Calcd for C₁₀H₁₇O₃I: C, 38.48; H, 5.49. Found: C, 38.62; H, 5.53.

2-(4-Nitrobenzyloxy-5-iodooxepane 9. In a round bottom flask equipped with a reflux condenser were placed compound **1** (115 mg, 0.45 mmol), 4-nitrobenzylic alcohol (1.5 equiv., 105.1 mg, 0.67 mmol), cyclohexane (25 mL), ether (4 mL), 3-tertbutyl-4-hydroxy-5-methylphenyl sulfide (0.2 equiv., 32.3 mg, 0.09 mmol), p-toluenesulfonic acid (0.25 equiv., 21.4 mg, 0.11 mmol). The mixture was stirred at reflux for 48 h. The organic layer was washed with a 0.1 M aqueous solution of sodium hydroxide and then with water, dried (sodium sulfate) and concentrated under vacuum. The crude product was purified by silica gel column chromatography (ether:pentane; 10:90) to give one isomer of product **9** (103 mg, 60% yield). ¹H NMR δ1.40 - 1.90 (m, 4H), 2.10 - 2.45 (m, 2H), 3.45 - 3.60 (m, 1H), 3.85 (ddd, J = 12.3, 9, 4.5 Hz, 1H), 4.20 (ddd, J = 11.2, 9, 3 Hz, 1H), 4.62 (d, J = 13.5 Hz, 1H), 4.84 (d, J = 13.5 Hz, 1H), 5.02 (d, J = 8 Hz, 1H), 7.58 (d, J = 9 Hz, 2H), 8.24 (d, J = 9 Hz, 2H). MS (DCI, NH₃) (m/z, %): 106 (100), 132 (16), 151 (13), 190 (12), 225 (22), 242 (41), 250 (11).

2-(Pro-2-enyl)oxepane 10.¹⁰ ¹H NMR δ 1.40 - 1.88 (m, 8H), 2.23 (m, 2H), 3.45 - 3.62 (m, 2H), 3.86 (dt, J = 11.8, 5.3 Hz, 1H), 5.00 - 5.14 (m, 2H), 5.85 (m, 1H). MS (DCI⁺, NH₃) (m/z, %): 141 (100), 158 (54).

3-Iodo-2-(prop-2-enyl)oxepane 11. This product was always obtained as a mixture with compound **10**. ¹H NMR δ 1.65 - 1.80 (m, 4H), 2.20 - 2.45 (m, 3H), 2.67 - 2.80 (m, 1H), 3.40 - 3.55 (m, 1H), 3.70 - 3.82 (m, 1H), 4.07 - 4.25 (m, 2H), 4.98 - 5.20 (m, 2H), 5.72 - 5.97 (m, 1H). MS EI (m/z, %): 41 (100), 69 (51), 85 (40), 98 (37), 139 (39), 225 (28).

3-Bromo-2-methoxyoxepane 13. The reaction conditions were similar to those used in the preparation of 3-iodo-2-methoxy oxepane 1 with HBB in place of HBI. Product 13 was obtained as a mixture of two diastereoisomers (70:30, 74% yield). ¹H NMR δ 1.45 - 2.30 (m, 6H), 3.42 (s, 2.19H, OMe, *trans*-isomer), 3.49 (s, 0.81H, OMe, *cis*-isomer), 3.50 - 4.05 (m, 2.73H), 4.20 (ddd, 0.27H, *cis*-isomer), 4.63 (d, J = 1.9 Hz, 0.27H, *cis*-isomer), 4.70 (d, J = 7.2 Hz, 0.73H, *trans*-isomer). ¹³C NMR δ (*cis-trans* mixture) 15.2, 22.8, 26.9, 29.46, 29.7, 33.8, 35.4, 54.5, 54.6, 55.6, 56.1, 63.9, 103.0, 108.1. MS EI (m/z, %): 41 (32.58), 61 (100), 69 (49). IR (film) (cm⁻¹): 2940, 1450, 1350, 1255, 1120, 1060.

3-Bromo-2-(4-nitrobenzyloxy)oxepane 14. In a round bottom flask with a reflux condenser were placed compound **13** (4.3 mmol, 894 mg), 4-nitrobenzylic alcohol (5.2 mmol, 786 mg), toluene (100 mL) and a catalytic amount of p-toluenesulfonic acid. The reaction mixture was stirred at reflux for one night. The organic layer was then washed with a 0.1 M aqueous solution of sodium hydroxide and water, dried (Na2SO4) and concentrated under vacuum. The crude product was purified by flash chromatography on silica gel (ether:pentane; 15:95 to 10:90) to lead to product **14** (3.85 mmol, 89.5% yield) as a mixture of two diastereoisomers (*trans:cis*; 85:15). *Trans*-isomer **14**. ¹H NMR δ 1.40 - 1.78 (m, 3H), 1.78 - 1.96 (m, 1H), 2.05 - 2.32 (m, 2H), 3.56 (ddd, J = 9.8, 8.4, 4.2 Hz, 1H), 3.87 (ddd, J = 11.2, 8.4, 4.2 Hz, 1H), 4.08 (ddd, J = 9.8, 6.8, 2.8 Hz, 1H), 4.66 (d, J = 13.5 Hz, 1H), 4.85 (d, J = 13.5 Hz, 1H), 4.89 (d, J = 6.8 Hz, 1H), 7.58 (d, J

= 8.4 Hz, 2H), 8.23 (d, J = 8.4 Hz, 2H). 13 C NMR δ 27.0, 29.6, 35.4, 54.1, 64.2, 68.3, 106.5, 123.5, 128.0, 145.2, 147.3. *Cis*-isomer **14**. 1 H NMR δ 1.50 - 2.00 (m, 4H), 2.05 - 2.22 (m, 1H), 2.22 - 2.40 (m, 1H), 3.70 (ddd, J = 11.2, 5.6, 4.2 Hz, 1H), 4.97 (ddd, J = 12.6, 8.4, 5.6 Hz, 1H); 4.25 (ddd, J = 8.4, 4, 1.4 Hz, 1H), 4.69 (d, J = 13.5 Hz, 1H), 4.85 (d, J = 1.4 Hz, 1H), 4.94 (d, J = 13.5 Hz, 1H), 7.60 (d, J = 8.4 Hz, 2H), 8.23 (d, J = 8.4 Hz, 2H). 13 C NMR δ 22.8, 29.1, 33.4, 54.2, 64.8, 68.2, 101.1, 123.3, 127.6, 145.1, 147.1. MS EI (m/z, %): 55 (30), 67 (40), 78 (39), 89 (47), 107 (23), 137 (100), 182 (32), 331 (0.57). IR (film) (cm⁻¹): 2940, 2870, 1610, 1530, 1350, 1110, 1060. Anal. Calcd. for C₁₃H₁₆BrNO₄: C, 47.29; H, 4.88. Found: C, 47.37; H, 4.75.

3-Bromo-2-(prop-2-enyl)oxepane 15. To a solution of compound **14** (2.66 mmol, 878.6 mg) in acetonitrile (6 mL) cooled at 0°C, under argon, were added allyltrimethylsilane (2.5 equiv., 6.65 mmol, 1.1 mL), BF₃.Et₃O (2.5 equiv., 6.65 mmol, 0.82 mL). After stirring at rt for one night, a saturated solution of sodium hydrogen carbonate was added. The aqueous layer was extracted with methylene chloride. The combined organic layers were dried (sodium sulfate) and concentrated under vacuum. The crude product was purified by flash chromatography on silica gel (ether:pentane; 10:90) to lead to product **15** (429 mg, 74% yield) as the sole isomer. ¹H NMR δ 1.60 - 1.90 (m, 4H), 2.08 - 2.40 (m, 2H), 2.60 - 2.75 (m, 2H), 3.40 - 3.55 (m, 1H), 3.64 (ddd, J = 9, 9, 3.9 Hz, 1H), 4.00 - 4.20 (m, 2H), 5.05 - 5.20 (m, 2H), 5.78 - 6.00 (m, 1H). ¹³C NMR δ 22.6, 30.5, 37.1, 38.5, 57.4, 72.8, 86.3, 117.3, 134.5. MS EI (m/z, %): 41 (100), 69 (77), 79 (20), 97 (13), 179 (40), 219 (0.19). Anal. Calcd. for C9H₁5BrO: C, 49.53; H, 6.93. Found: C, 49.67; H, 7.00.

3-Hydroxy-2-(prop-2-enyl)oxepane 16. Compound **15** (1.54 mmol, 337.5 mg) was placed in a round bottom flask with nitromethane (17 mL) and phosphate buffer pH 7 (8 mL). The solution was cooled at 0°C. After addition of silver trifluoroacetate (2 equiv., 3.08 mmol, 680 mg), the reaction mixture was stirred at rt for one night, and then filtered. Azeotropic removal of solvent using acetone was pursued before purification by flash chromatography on silica gel (ether:pentane; 1:1). The allylic alcohol **16** was obtained as a mixture of two diastereoisomers (*trans:cis*; 97:3, measured by GC analysis) (210.2 mg, 87.5% yield). *Trans*-isomer **16**. ¹H NMR δ 1.45 - 2.05 (m, 7H), 2.20 - 2.39 (m, 1H), 2.42 - 2.55 (m, 1H), 3.28 (ddd, J = 7.9, 7.9, 3.9 Hz, 1H), 3.45 - 3.60 (m, 1H), 3.60 - 3.72 (m, 1H), 3.95 - 4.10 (m, 1H), 5.05 - 5.20 (m, 2H), 5.85 - 6.05 (m, 1H). ¹³C NMR δ 20.4, 30.5, 35.8, 38.5, 71.2, 75.0, 84.6, 116.7, 135.3. MS EI (*m/z*, %): 41 (36), 57 (100), 71 (14), 84 (15), 115 (27), 156 (0.06). IR (film) (cm⁻¹): 3620, 2940, 2870, 1645, 1605, 1445, 1110. Anal. Calcd. for C9H₁₆O₂: C, 69.18; H, 10.33. Found: C, 69.35; H, 10.47.

Ozonolyis of compound 16. The reaction conditions applied to the preparation of the allylic alcohol 16 (1.35 mmol, 210 mg) were those used for preparation of compounds 3a and 3b. The crude product 17 was used for the next reaction without purification.

3-Hydroxy-2-(3-methoxyprop-2-enyl)oxepane 18. To a solution of methoxymethyltriphenylphosphonium chloride (3 equiv., 4.05 mmol, 1.38 g) in dry THF (6 mL) under argon at -78°C, was added phenyllithium (3 equiv., 4.05 mmol, 2 mL of 2 M solution in ether). The solution was stirred for 15 min. at -78°C and then the crude product **17** was added in solution in THF (3 mL). The flask was slowly warmed to rt and then stirred 24 h at this temperature. Water and dichloromethane were added and aqueous layer was extracted with methylene chloride. The combined organic layers were dried (Na₂SO₄) and concentrated under vacuum. Purification of the crude product by flash chromatography (ether:pentane; 30:70) gave compound **18** (113 mg, 45% yield from **16**) as a mixture of two isomers (*E*:*Z*; 70:30). ¹H NMR δ1.42 - 1.85 (m, 4H), 1.9 - 2.05 (m, 1.3H), 2.05 - 2.22 (m, 1H), 2.50 - 2.40 (m, 1H), 2.43 - 2.55 (m, 0.7H), 3.60 (td, J = 7.9, 4 Hz, 0.7H), 3.28 (dt, J = 10.6, 5.3 Hz, 0.3H), 3.45 - 3.74 (m, 2H), 3.53 (s, 2.1H), 3.63 (s, 0.9H), 3.84 - 4.07 (m, 1H), 4.55 (dt, J = 9.2, 6.6 Hz, 0.3H), 4.83 (dt, J = 11.9, 6.6 Hz, 0.7H), 6.04 (d, J = 6.6 Hz, 0.3H), 6.40 (d, J = 11.9 Hz, 0.7H). IR (film) (cm⁻¹): 3400, 2920, 2850, 1650, 1450, 1205, 1100. Anal. Calcd. for C10H18O3: C, 64.47; H, 9.75. Found: C, 64.60; H, 9.66.

(1S*,6R*)-4-Bromo-3-methoxy-2.7-dioxabicyclo[5.4.0]undecane 19. The reaction conditions were those used for the preparation of compound 13. Starting from 18 (0.61 mmol, 113 mg), we obtained product 19 (50

mg, 31% yield) after flash chromatography on silica gel (ether:pentane; 10:90) as a mixture of three diastereoisomers. First eluted fraction, 2 isomers (80:20). 1 H NMR δ 1.40 - 2.40 (m, 8H), 3.10 - 3.30 (m, 0.4H), 3.36 (s, 2.4H), 3.43 (s, 0.6H), 3.46 - 3.68 (m, 1.6H), 3.68 - 3.90 (m, 2H), 3.90 - 4.10 (m, 0.2H), 4.20 (t, J = 3.3 Hz, 0.8H), 4.62 (d, J = 3 Hz, 0.2H), 4.72 (s, 0.8H). 13 C NMR δ major isomer 20.1, 28.8, 34.4, 34.5, 48.3, 54.8, 68.9, 73.0, 73.3, 99.3. minor isomer 20.0, 28.9, 29.7, 34.2, 36.4, 44.8, 55.3, 71.9, 77.3, 97.7. MS EI (m/z, %): 41 (35), 57 (21), 67 (23), 85 (100), 98 (45), 265 (0.08). IR (film) (cm⁻¹): 2950, 2880, 1450, 1140, 1030. Anal. Calcd. for C₁₀H₁₇O₃Br: C, 45.30; H, 6.46. Found: C, 45.62; H, 6.68. Second eluted fraction, 1 isomer. 1 H NMR δ 1.40 - 2.30 (m, 7H), 2.40 (dt, J = 12.7, 4.2 Hz, 1H), 3.15 - 3.50 (m, 2H), 3.54 (s, 3H), 3.62-3.93 (m, 3H), 4.32 (d, J = 8.5 Hz, 1H). 13 C NMR δ 19.8, 29.1, 34.0, 41.4, 46.3, 57.1, 69.1, 77.2, 79.8, 104.8. MS EI (m/z, %): 41 (34), 57 (23), 67 (23), 85 (74), 98 (100), 265 (0.30). IR (film) (cm⁻¹): 2950, 2880, 1460, 1125, 1060. Anal. Calcd. for C₁₀H₁₇O₃Br: C, 45.30; H, 6.46. Found: C, 45.51; H, 6.53.

REFERENCES

- 1. Yasumoto, T.; Murata, M. Chem. Rev. 1993, 93, 1897.
- Alvarez, E.; Candenas, M.-L.; Pérez, R.; Ravelo, J.L.; Martin, J.D. Chem. Rev. 1995, 95, 1953. Mori, Y. Chem. Eur. J. 1997, 3, 849.
- For some recent results see: Oishi, T.; Nagumo, Y.; Hirama, M. J. C. S., Chem. Comm. 1998, 1041.
 Mori, Y.; Yaegashi, K.; Furukawa, H. Tetrahedron 1997, 53, 12917. Gleason, M.M.; McDonald, F.E. J. Org. Chem. 1997, 62, 6432. Clark, J.S.; Kettle, J.G. Tetrahedron Lett. 1997, 38, 127. Atsuta, H.; Fujiwara, K.; Murai, A. Synlett. 1997, 307. Alvarez, E.; Pérez, R.; Rico, M.; Rodriguez, M. R.; Martin, J. D. J. Org. Chem. 1996, 61, 3003.
- 4. Hemibrevetoxin B: Nicolaou, K.C.; Reddy, K.R.; Skokotos, G.; Sato, F.; Xiao, X.-Y.; Hwang, C.K. J. Am. Chem. Soc. 1993, 115, 3558. Kadota, I.; Park, J.Y.; Koumura, N.; Polland, G.; Matsokawa, Y.; Yamamoto, Y. Tetrahedron Lett. 1995, 36, 5777. Ishihara, J.; Murai, A. Synlett 1996, 363. Morimoto, M.; Matsukura, H.; Nakata, T. Tetrahedron Lett. 1996, 37, 6365. Mori, Y.; Yaegashi, K. Furukawa, H. J. Am. Chem. Soc. 1997, 119, 4557.
 - Brevetoxin B: Nicolaou, K.C.; Theodorakis, E.E.; Rutjes, F.P.J.T.; Tiebes, J.; Sato, M.; Untersteller, E.; Xiao, X.-Y. J. Am. Chem. Soc. 1995, 117, 1171. Nicolaou, K.C.; Rutjes, F.P.J.T.; Theodorakis, E.A.; Tiebes, J.; Sato, M.; Untersteller, E. J. Am. Chem. Soc. 1995, 117, 1173. See: Nicolaou, K.C. Angew. Chem. Int. Ed. Engl. 1996, 35, 589.
- Suzuki, K.; Mukaiyama, T. Chem. Lett. 1982, 683 and 1525. Ting, P. C.; Bartlett, P. A. J. Am. Chem. Soc. 1984, 106, 2668. Paquet, F.; Sinaÿ, P. Tetrahedron Lett. 1984, 25, 3071. Hamada, Y.; Kawai, A.; Shioiri, T. Tetrahedron Lett. 1984, 25, 5413. Doyle, M. P.; Griffin, J.H., Chinn, M. S.; Van Leusen, D. J. Org. Chem. 1984, 49, 1917. Ha, D.-C.; Hart, D. J. Tetrahedron Lett. 1987, 28, 4489. Kerwin, S.M.; Paul, A.G.; Heathcock, C.H. J. Org. Chem. 1987, 52, 1686. Koft, E. R.; Dorff, P.; Kullnig, R. J. Org. Chem. 1989, 54, 2936. Fabre, V.; Lila, C.; Saroli, A.; Doutheau, A. Tetrahedron 1989, 45, 7765. Vlahov, I. R.; Vlahova, P. I.; Schmidt, R. R. Tetrahedron Lett. 1992, 33, 7503. Maudrin, J.; Barrère, B.; Chantegrel, B.; Deshayes, C.; Quash, G.; Doutheau, A. Bull. Soc. Chim. Fr. 1994, 131, 400. Devasagayaraj, A.; Schwink, L.; Knochel, P. J. Org. Chem. 1995, 60, 3311.
- 6. Brunel, Y.; Rousseau, G. J. Org. Chem. 1996, 61, 5793.
- 7. Sweet, F.; Brown, R.K. Can. J. Chem. 1966, 44, 1571.
- 8. Fleming, I.; Dunogués, J.; Smithers, R. Organic Reactions, 1989, 37, 57.
- 9. Hunter, R.; Michael, J.P.; Tomlinson, G.D. Tetrahedron, 1994, 50, 871.
- 10. Nicolaou, K.C.; McGarry, D.G.; Somers, P.K.; Ogilvie, W.W.; Yiannikouros, G.; Prasad, C.V.C.; Veale, C.A.; Hark, R.R. J. Am. Chem. Soc. 1990, 112, 6263.